MOLECULAR STRUCTURE AND ABSOLUTE CONFIGURATION OF 2,3-DIDEHYDRO-1β,10β-EPOXYPSILOSTACHYIN B (ISOPAULITIN)

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ABSTRACT.—The molecular structure, conformational analysis, and absolute configuration of isopaulitin [1], a sesquiterpene lactone, have been determined by single-crystal X-ray diffraction analysis. The correlation between the position and stereochemistry of the γ -lactone fusion to the cycloheptane ring and the sign of the Cotton effect of the n- π * transition of the C=C-C=O chromophore in 1 has been analyzed.

Isopaulitin [1] has been isolated from leaves of Ambrosia cumanensis HBK. (Compositae) (EtOH extract), a medicinal plant known as "altamisa" in El Salvador, where it was collected, and which has antirheumatic, vermifuge, and antitumor properties (1). An X-ray diffraction study of 1 was undertaken to determine its absolute configuration, which will be used as reference data to compare the stereochemistry in related psilostachyinolides such as altamisin, paulitin, and lagosin, other sesquiterpene dilactones isolated from A. cumanensis (1-4).

Isopaulitin [1] [mp 147–150°, [α]²⁰D -20° (CHCl₃)] has a molecular formula of C₁₅H₁₆O₅, as determined by elemental analysis and mass spectroscopy. The compound showed a uv band consistent with the presence of an α , β' -unsaturated- γ lactone (λ max 225 nm, ϵ 23500). Evi-

dence of γ -lactone and δ -lactone rings in the molecule of 1 was obtained from the absorption bands in the ir spectrum at 1770 cm⁻¹ and at 1725 cm⁻¹, respectively. The ¹H-nmr spectrum recorded in CDCl₃ provided further information about the α,β' -unsaturated γ -lactone of the psilostachyinolide type (5,6), e.g., the signals at δ 4.76 (1H, d, J=9 Hz) for H-6 which suggested a cis-fused five membered ring; at δ 5.54 and 6.22 as two doublets, J=3 Hz for the exocyclic methylene at C-11; the AB system at δ 6.28 (1H, d, J=10 Hz) and 6.39(1H, d, J=10Hz) for the δ -lactone conjugated olefin protons; and, finally, two tertiary methyl signals at δ 1.46 and 1.59 for the methyls at C-5 and C-10, respectively. The 13 Cnmr spectrum of 1 in CDCl₃ exhibited two signals at δ 61.9 and 70.9 for the C-1 and C-10 carbons, respectively, of the



Structure of Isopaulitin

oxacyclopropane ring. The former is normal for a β -carbon atom relative to the methyl group at C-5 (7).

Crystallographic analysis of **1** was carried out to determine unambiguously its absolute configuration and by inference, those of biogenetically related compounds (4,7). Crystals of **1** are orthorhombic, space group P2₁2₁2₁, with a=13.918(1), b=10.526(1), c=9.220(1)Å, and V=1350.7(2) Å³, using CuK α radiation (λ =1.5418 Å). The molecular formula is C₁₅H₁₆O₅, mol wt 276.29, D_c=1.359(2)gcm⁻³,Z=4,F(000)=586. Correction by absorption was neglected, μ =8.11 cm⁻¹.

Fractional atomic co-ordinates and equivalent temperature factors are given in Table 1; bond lengths and angles are given in Tables 2 and 3, respectively; and tion, absolute configuration, and numbering scheme are shown in Figure 1.¹

From the values of the torsion angles (Table 4), it may be interpreted that the cycloheptane ring conformation is closer to the D_2 twist-chair than to the D_2 chair form, and the asymmetric parameters (8) show that the dominant symmetry is the rotational one D₂ through C-7. The C-C bond distances and angles for this ring (Tables 2 and 3) show the expected values for carbon-oxygen bonds and C-C bonds, the last ranging within 1.50 to 1.54 Å, except for C-1–C-2=1.477(3) which agrees well with a $C-sp^3-C-sp^2$ bond and C-1-C-5=1.552(3), probably due to steric hindrance (9) between C-14 and the oxygen of the oxacyclopropane ring. C-2-C-3=1.317(3) shows a short doublebond value. The bond distances and angles

Atom	x/a	у/b	z/c	Ueq	
0-1	8471 (1)	8778 (1)	8678 (2)	52 (0)	
O-2	9995 (1)	8979 (2)	9195 (2)	74 (0)	
O-3	7192 (1)	6048 (1)	6860 (1)	52 (0)	
O-4	5919(1)	10794 (2)	10406 (2)	76 (0)	
O-5	6985 (1)	10192 (1)	8735 (2)	54 (0)	
C-1	7865 (1)	7099 (2)	6994 (2)	41 (0)	
C-2	8853 (2)	6598 (2)	7117 (2)	50 (0)	
C-3	9538 (2)	7150 (2)	7875 (2)	54 (0)	
C-4	9378 (2)	8348 (2)	8620 (2)	52 (0)	
C-5	7607 (1)	8112 (2)	8153 (2)	41 (0)	
C-6	7046 (1)	9265 (2)	7563 (2)	42 (0)	
C- 7	5997 (1)	9120 (2)	7062 (2)	43 (0)	
C-8	5723 (1)	7891 (2)	6323 (2)	48 (0)	
C-9	6430 (2)	7544 (2)	5129 (2)	50 (0)	
C-10	7329(1)	6860 (2)	5608 (2)	46 (0)	
C-11	5443 (1)	9469 (2)	8394 (2)	46 (0)	
C-12	6089 (2)	10215 (2)	9310 (2)	52 (0)	
C-13	4547 (2)	9240 (2)	8756 (3)	58 (1)	
C-14	7162 (2)	7490 (2)	9476 (2)	52 (1)	
C-15	7858 (2)	6288 (3)	4332 (3)	66 (1)	

TABLE 1. Atomic Parameters $(\times 10^4)$ for $\mathbf{1}$ [C₁₅H₁₆O₅ and equivalent temperature factors (\mathbf{A}^2) ($\times 10^3$) with e.s.d.'s in parentheses.⁴]

 $^{a}\text{Ueq}=(1/3)\boldsymbol{\Sigma}_{i}\boldsymbol{\Sigma}_{j}\boldsymbol{U}_{ij}\boldsymbol{a}_{i}^{*}\boldsymbol{a}_{j}^{*}\boldsymbol{a}_{j}\boldsymbol{a}_{j}\cos(\boldsymbol{a}_{i}\boldsymbol{a}_{j}).$

torsion angles are given in Table 4. The X-ray analysis confirms that **1** is a sesquiterpene lactone wherein the δ -lactone and the α -methylene- γ -lactone rings are cis-fused to a cycloheptaneoxacyclopropane bicyclic system whose constitu-

¹Hydrogen coordinates, thermal parameters, bond distances and angles, and observed and calculated structure factors have been deposited with the Cambridge Crystallographic Data Centre and can be obtained upon request from Dr. Olga Kennard, University Chemical Laboratory, 12 Union Road, Cambridge CB2 1EZ, UK.

1.342 (3)	C-3-C-4	1.453 (3)
1.474 (2)	C-5C-6	1.542 (3)
1.208 (3)	C-5-C-14	1.517 (3)
1.455 (2)	C-6-C-7	1.539 (2)
1.449 (2)	C-7–C-8	1.511 (3)
1.204 (3)	C-7–C-11	1.496 (2)
1.458 (3)	C-8-C-9	1.521 (3)
1.355 (3)	C-9–C-10	1.510 (3)
1.477 (3)	C-10-C-15	1.513 (3)
1.552 (3)	C-11-C-12	1.462 (3)
1.501 (2)	C-11-C-13	1.313 (3)
1.317 (3)		
	1.342 (3) 1.474 (2) 1.208 (3) 1.455 (2) 1.449 (2) 1.204 (3) 1.455 (3) 1.455 (3) 1.455 (3) 1.455 (3) 1.552 (3) 1.552 (3) 1.501 (2) 1.317 (3)	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$

TABLE 2. Bond Lengths (Å) for $1 [C_{15}H_{16}O_5 (e.s.d.'s in parentheses)]$.

for the oxacyclopropane ring show the geometrically expected values for a tetrasubstituted derivative: thus, C-1–C-10=1.501(2), C-1–O-3=1.455(2), and C-10–O-3=1.449(2) Å, while the angles C-1–O-3–C-10=62.3(1), O-3–C-10=58.7(1), and O-3–C-10–C-1=59.1(1)Å.

The δ -lactone ring shows a distorted boat conformation and the conjugate system C-2=C-3-C-4=O-2 is planar, with a maximum deviation of 0.053(2) Å, as is the olefinic double bond plane C-1-C-2=C-3-C-4, with the maximum deviation of 0.011(3) Å. The dihedral angle between both planes is 3°. This conformation is observed in a similar compound (10). The torsion angles in the cycloheptane ring show a C_2 axis of symmetry and the Σ_2 calculation for the dihedral angles (8) is approximately 0°, which is in agreement with the twist-chair stable conformation for cycloheptane rings (11,12). The α -methylene- γ -lactone ring adopts an envelope conformation and is cis-fused at C-6–C-7 to the cycloheptane ring.

A number of terpenoids containing the α,β -methylene- γ -lactone ring have been subjected to X-ray examination and the results show that the most of these lactone rings deviate significantly from the ideal envelope conformation (10).

The γ -lactone ring in **1** has the C=O and the exocyclic methylene groups slightly deviating from coplanarity. More-

C-4-O-1-C-5	O-5-C-6-C-5 107.1 (1)
C-1–O-3–C-10	O-5-C-6-C-7 103.5 (1)
C-6–O-5–C-12	C-5-C-6-C-7 120.5 (2)
O-3–C-1–C-2	C-6–C-7–C-8 117.4 (2)
O-3-C-1-C-5	C-6-C-7-C-11 102.6 (1)
O-3-C-1-C-10	C-8-C-7-C-11 116.8 (2)
C-2–C-1–C-5	C-7-C-8-C-9 111.6 (2)
C-2C-1C-10	C-8-C-9-C-10 116.0 (2)
C-5–C-1–C-10	O-3-C-10-C-1
C-1–C-2–C-3	O-3-C-10-C-9 113.9 (2)
C-2–C-3–C-4	O-3-C-10-C-15 116.7 (2)
O-1–C-4–O-2	C-1-C-10-C-9 125.5 (2)
O-1–C-4–C-3	C-1–C-10–C-15 119.1 (2)
O-2-C-4-C-3	C-9–C-10–C-15 111.5 (2)
O-1–C-5–C-1	C-7–C-11–C-12 106.8 (2)
O-1–C-5–C-6	C-7–C-11–C-13 130.8 (2)
O-1–C-5–C-14 105.9 (2)	C-12–C-11–C-13 122.4 (2)
C-1–C-5–C-6	O-4-C-12-O-5 121.2 (2)
C-1–C-5–C-14 110.6 (2)	O-4-C-12-C-11 129.5 (2)
C-6–C-5–C-14114.6 (2)	O-5-C-12-C-11 109.3 (2)

TABLE 3. Bond Angles (°) for $1 [C_{15}H_{16}O_5 (e.s.d.'s in parentheses)]$.

O-1-C-4-C-3-C-2	C-1-C-10-C-9-C-836.2 (3)
O-1–C-5–O-1–O-3 154.4 (1)	C-2-C-1-O-3-C-10111.5 (2)
O-1-C-5-C-1-C-2	C-2-C-1-C-5-C-6 137.3 (2)
O-1-C-5-C-1-C-10137.0 (2)	C-2-C-1-C-5-C-1491.4 (2)
O-1C-5C-6O-553.6 (2)	C-2-C-1-C-10-C-9164.3 (2)
O-1–C-5–C-6–C-7 –171.4 (2)	C-2-C-1-C-10-C-158.4 (3)
O-2-C-4-O-1-C-5174.8 (2)	C-3-C-2-C-1-C-516.2 (3)
O-2-C-4-C-3-C-2171.2 (2)	C-3-C-2-C-1-C-10 148.3 (2)
O-3-C-1-C-2-C-3147.5 (2)	C-3-C-4-O-1-C-5 4.8 (3)
O-3-C-1-C-5-C-6	C-4-O-1-C-5-C-6143.1 (2)
O-3-C-1-C-5-C-14	C-4-O-1-C-5-C-14 98.0 (2)
O-3-C-1-C-10-C-9	C-5-C-1-O-3-C-10 118.0 (2)
O-3-C-1-C-10-C-15105.4 (2)	C-5-C-1-C-10-C-91.8 (3)
O-3-C-10-C-1-C-2	C-5-C-1-C-10-C-15 154.0 (2)
O-3-C-10-C-1-C-5100.6 (2)	C-5-C-6-O-5-C-12105.4 (2)
O-3-C-10-C-9-C-8	C-5-C-6-C-7-C-835.8 (2)
O-4-C-12-O-5-C-6 170.4 (2)	C-5-C-6-C-7-C-11 93.7 (2)
O-4-C-12-C-11-C-7 171.8 (2)	C-6-O-5-C-12-C-1110.0 (2)
O-4-C-12-C-11-C-136.3 (4)	C-6-C-5-C-1-C-1025.8 (2)
O-5-C-6-C-5-C-1172.1 (1)	C-6-C-7-C-8-C-9 $-48.9(2)$
O-5-C-6-C-5-C-14	C-6-C-7-C-11-C-12 20.9 (2)
O-5-C-6-C-7-C-8155.4 (2)	C-6-C-7-C-11-C-13161.4 (2)
O-5-C-6-C-7-C-1125.9 (2)	C-7-C-6-O-5-C-12 23.0 (2)
O-5-C-12-C-11-C-77.8 (2)	C-7-C-6-C-5-C-1459.2 (2)
O-5-C-12-C-11-C-13 174.2 (2)	C-7-C-8-C-9-C-10 83.5 (2)
C-1-O-3-C-10-C-9118.4 (2)	C-8-C-7-C-11-C-12 150.7 (2)
C-1–O-3–C-10–C-15 109.5 (2)	C-8-C-7-C-11-C-1331.5 (3)
C-1-C-2-C-3-C-4	C-8-C-9-C-10-C-15 166.4 (2)
C-1-C-5-O-1-C-422.3 (2)	C-9-C-8-C-7-C-11171.4 (2)
C-1-C-5-C-6-C-7 70.1 (2)	C-10-C-1-C-5-C-14 105.5 (2)

TABLE 4. Dihedral Angles (°) for $1 [C_{15}H_{16}O_5 (e.s.d.'s in parentheses)]$.

over, the related bond lengths C-11–C-13=1.313(3), C-12–O-4=1.204(3), C-11–C-12=1.462(3), and O-5–C-12=1.355(3) confirm the absence of conjugation through C=O and the double bond. In contrast, the C=O and the C-2=C-3 double bond groups in the δ -lactone ring show some con-

jugation effect.

From measurements of the sign of the Cotton effect of the $n-\pi^*$ transition of the α -methylene- γ -lactone chromophore in a number of sesquiterpene lactones (13), it has been established that, with some exceptions, the C-6–C-7 cis-fused lactones give a positive Cotton effect,



FIGURE 1. Molecule of isopaulitin [1] showing the atomic labeling.

whereas a negative Cotton effect is found when the γ -lactone is trans-fused across the C-6-C-7 bond. On the other hand, the sign of the Cotton effect of the α methylene-y-lactone chromophore has been related (11) to the chirality of the C=C-C=O group which permits determination of the absolute configuration of the molecule. For 1 with a cis-fused C-6-C-7 lactone ring the experimentally determined Cotton effect is positive, λ max (EtOH) (log ϵ) 262 nm (1.3), which is in agreement with the above predictions. Moreover, the torsion angle of the C=C-C=O group is -6° , which is paired with that of the endocyclic torsion angle about the C-6–C-7 bond (-26°) and with the latter angle about three times greater than the former. All these results are in good agreement with the correlations established.

Figure 1 shows the absolute configuration of isopaulitin [1] which was determined by comparing the more relevant Bijvöet pairs. The absolute configurations of the five chiral centers in isopaulitin are given by the torsion angles, being: C-1, R, C-5, S, C-6, R, C-7, S, and C-10, S.

EXPERIMENTAL

PLANT MATERIAL.—Isopaulitin [1] has been isolated from an extract of the leaves of *Ambrosia cumanensis* H.B.K., a species which belongs to the tribe Heliantheae (Compositae). It is known as "altamisa" in the Republic of El Salvador, where the plant was collected (7).

X-RAY CRYSTALLOGRAPHIC ANALYSIS OF analysis were obtained from slow evaporation of a dilute EtOAc solution. A crystal of $0.20 \times 0.16 \times 0.30$ mm dimensions was used for the crystallographic study. Accurate cell dimensions were determined by least-squares analysis of setting angles of 34 reflections ($10 < \theta < 32^\circ$), using graphite monochromated CuKa radiation automatically located and centered on a Philips PW 1100 diffractometer. Two reflections were measured every 90 reflections to ascertain crystal stability, and no significant variation was observed. All the reflections were corrected for Lorentz and polarization effects, but no absorption correction was applied. For the intensity meaasurement, reflections were surveyed in the range $2\theta \leq 65$, from the 2292 reflections measured as independent Friedel pairs, 2080 were considered as observed, satisfying the criterion $I \ge 2\sigma(I)$, and were used in the subsequent calculations.

The crystal structure was partially elucidated by direct methods (15). After preliminary adjustment of the positions of the C and O atoms, the Hatoms were positioned from difference Fourier maps. Thereafter, several cycles of full-matrix leastsquares calculations were carried out with anisotropic thermal parameters for C and O atoms and the H atoms were included as fixed contributors, and convergence was reached at R=0.044 and R_w =0.048 with a weighting scheme (16) to prevent trends in w $\Delta^2 F$ vs $< F_o >$ and vs $< \sin\theta/\lambda >$ including both \pm hkl pairs. The final difference synthesis showed no peaks exceeding 0.40 eÅ⁻³, and the value of the parameters shift error was less than 0.01.

The absolute configuration (17) was determined by comparison of the 76 more relevant Bijvöet pairs, considering reflections with $F_o > 10\sigma(F_o)$, with $\Delta F_c > 0.07$. The following discrepancy indexes were found: $R_1 = \Sigma([F_o(h)] - [F_o(-h)]) - ([F_c(h)] - ([F_c(-h)]) - N = 0.219 (0.280 for the reversed enantiomer); <math>R_2 = 1 + \Sigma[(R_o - R_c) - 1/N] = 1.031 (1.035$ for the reversed enantiomer); $R_3 = \Sigma[(\Delta I_o - \Delta I_c) / \Sigma(\Delta I_o)] = 0.916 (1.242$ for the reversed enantiomer); where $R_o = [F_o(h) / [F_c(-h)]$, $R_c = [F_c(h) / [F_c(-h)]$. $\Delta I_o = [F_o(h)]^2 - [F_o(-h)]^2$, $\Delta I_c = [F_c(h)]^2 - [F_c(-h)]^2$.

The atomic scattering factors and the anomalous dispersion correction were taken from the literature (18).

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